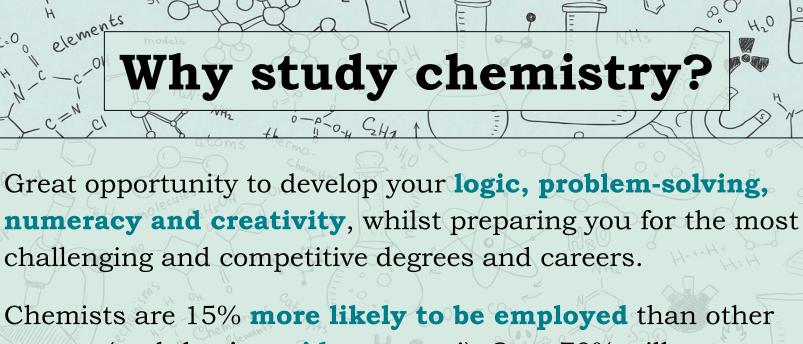


Why study chemistry?

One of the most **respected** and **impressive** A-Levels to have, in both science and non-science related careers.

Chemistry plays a role in almost every action on earth, in every object we touch. If you're interested in **understanding**, **contributing to and improving the world around you**, then Chemistry will open doors you may not have even thought of.

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Chemists are 15% more likely to be employed than other careers (and they're paid more too!). Over 70% will enter a professional career.

Structure

CHC+4H2

ON

The A-Level Chemist at RPS Organised - keep on top of your notes Motivated to study outside of lessons Ask for help when you need it

- Revise for every test as if it counts towards your final grade it's the only way to make progress
- **Resilient** Chemistry is not easy and you won't always hit your target grade on the first go at a topic

I CoHc+4H2

• In it for the long haul - it will take the whole 2 years of hard work, to get to the grade you'd like

BUT IT IS SO, SO WORTH IT

Where can Chemistry take you? Medicine Sales Forensics Auditor Medicinal Computing Cosmetics niroscope Web design Environment Consultancy Food industry Armed services Biotechnology Chemistry CaH12 Non profit Sector Nanotechnology Science writing Analytic services Management Further Education Ca-2ē = Ca Accountancy Academia Publishing Reseach Marketing Tutoring 0,1 Policy

OCR Chemistry A Year 12 Modules

atoms

structure

CcHc+4H2

- Module 3 Periodic table and energy
- The periodic table and periodicity
- Group 2 and the halogens
- Qualitative analysis
- Enthalpy changes

omic Structure Reaction rates and equilibrium (qualitative)

Module 2 – Foundations in chemistry

Diquid

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- Atoms, compounds, molecules and equations Amount of substance
- Acid-base and redox reactions
- Electrons, bonding and structure

- Basic concepts
 - **Hydrocarbons**

Module 4 - Core organic chemistry

- Alcohols and haloalkanes
- Organic synthesis
- Analytical techniques (IR and MS)



Module 5 – Physical chemistry and transition elements

Reaction rates and equilibrium (quantitative)

Enthalpy, entropy and free energy

Structure

C.Hc+4H2 (

- pH and buffers
- Redox and electrode potentials
- nedox dila electrode potentials
- Transition elements

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Module 6 – Organic chemistry and analysis

NH3

H20

- Aromatic compounds
- Carbonyl compounds
- Carboxylic acids and esters
- Nitrogen compounds
 - Organic synthesis

Polymers

Chromatography and spectroscopy (NMR)

